# A Convenient Synthesis of 2*H*-1,4-Benzoxazines, 3*H*-Indol-3-ones, and 2,3-Dihydrobenzoxazoles

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(Received March 18, 2002)

3,5-Di-*tert*-butyl-1,2-benzoquinone 1-oxime (**1b**) reacted with ester ylides **2a,b** to give the corresponding 2*H*-1,4-benzoxazin-2-ones **9a,b** along with the 3*H*-indol-3-one **11**, whereas with keto ylides **12a,b**, 2,3-dihydrobenzoxazoles **13a,b** were isolated. Conversely, the reaction of **1b** with moderated phosphonium salt **15** proceeded under phase-transfer catalysis conditions to afford the 2*H*-1,4-benzoxazine derivative **19**, instead of the expected Wittig reaction product.

We reported earlier<sup>1</sup> that 3,5-di-*tert*-butyl-1,2-benzoquinone reacted with trialkyl phosphites to give pentaoxyphosphoranes, as presumably observed with *o*-quinones, whereas with dialkyl phosphonates an anomalous behavior was shown, whereupon a ring attack occurred to give phosphonate adducts. We also found that the dione system in the quinone in question behaved differently toward Wittig<sup>2a,b</sup> and Wittig-Horner reagents.<sup>2c</sup>

Later on, in a continuing exploration of the attack of alkylidenephosphoranes on carbon–nitrogen systems,<sup>3</sup> we described<sup>3g</sup> the synthesis of substituted 2,3-dihydro-1H-indazoles **4a,b**, 2H-1,4-benzoxazin-2-one **5** and 1,2-dihydrocinnoline **6** by applying phosphorus ylides **2** or **3** on 3,5-di-*tert*-butyl-1,2-benzoquinone 1-phenylhydrazone (**1a**) (e.g. Scheme 1). It has been pointed out that the substitution pattern in **1** is such as to obstruct (for steric hindrance reasons) a nucleophilic approach by a ylide phosphorane to the aryl-carbonyl in **1a**. The effect of the neighboring t-Bu moiety on the C–2(O) group would be expected to be quite unfavorable.

As a sequel, the work detailed herein describes the reactions of 3,5-di-*tert*-butyl-1,2-benzoquinone 1-oxime (**1b**) with stabilized ester-**2a,b** or keto-**12a,b** ylide phosphoranes and a moderate allyltriphenylphosphonium bromide **15**. The methodology has led to a facile synthesis of the title compounds, ben-

zo[1,2-*b*]-fused *N*-heterocycles, which constitute an important class of compounds, many of which exhibit significant pharmaceutical and biological potency.<sup>4,5</sup>

#### **Results and Discussion**

The required o-quinone monoxime 1b was easily obtained in a reasonable yield (48%) from a reaction of the parent oquinone with hydroxylamine hydrochloride in boiling ethyl alcohol for 24 h. The treatment of **1b** with (ethoxycarbonylmethylene)triphenylphosphorane (2a, 2 molar amounts) in boiling chloroform (or in toluene; best yield in chloroform) gave ethyl 6,8-di-*tert*-butyl-2-oxo-2*H*-1,4-benzoxazine-3-acetate (9a) (41%) along with unexpected 4,6-di-tert-butyl-7-hydroxy-3Hindol-3-one (11) (28%) (Scheme 2). A similar treatment of 1b with (methoxycarbonylmethylene)triphenylphosphorane (2b) led to 2H-1,4-benzoxazin-2-one **9b** and 3H-indol-3-one **11** in 44 and 22% yields, respectively. The <sup>1</sup>H and <sup>13</sup>C NMR spectra of 9b are similar to those of 9a, except for the ester group, which displays characteristic resonances with appropriate chemical shifts. Repetition of the reaction between equimolar amounts of 1 and 2a,b again afforded 9a,b and 11 in addition to the unchanged 1b. The structures 9 and 11 were assigned based on elemental analyses and spectral data. Thus, 2H-1,4-

Scheme 2.

1b + 
$$Ph_3P = CHCOR^1$$
12a,  $R^1 = Ph$ 
b,  $R^1 = Me$ 
13a,b
14a,b

13 and 14, R1 as in 12

Scheme 3.

benzoxazin-2-ones **9a,b** showed strong absorptions at 1763 (**9a**) and 1772 cm<sup>-1</sup> (**9b**), characteristic of the carbonyl groups in similar six-membered heterocyclic compounds (cf.  $\delta$  lactones<sup>6</sup>). Other bands appeared at ~1720 [C(O), ester] and ~1590 cm<sup>-1</sup> (C=N). The <sup>1</sup>H NMR spectra of compounds **9a,b** exhibit a singlet for their C-9-methylene protons around  $\delta$  2.6, in agreement with the suggested structure, and rule out the alternative arylidene form. Furthermore, the distinguishing features of the <sup>13</sup>C NMR spectra of **9a,b** were the presence of signals at  $\delta$  ~28 (CH<sub>2</sub>COR), ~160 [C(O), ester] and at ~172 [C-1-(O)].

The respective mechanism for the formation of 9 might involve an initial Wittig olefination of 1b in its tautomeric nitroso form 1bA, leading to intermediate 7, followed by the addition of a second ylide species, 2a or 2b. The further elimination of triphenylphosphine and  $\delta$ -lactonization of the thusformed intermediate 8 yielded the final products 9a, b accompanied by extrusion of an alcoholic moiety (Scheme 2-i). Meanwhile, according to a mechanism suggested by the referee, and outlined in Scheme 2-ii, the formation of compound 11 involved the intermediate 10, which arose from an intramolecular Friedel–Crafts type condensation of intermediate 7.8 Further elimination of the appropriate alcoholic moiety from 10

afforded **11**. However, the latter step occurred through a carbabion mechanism, driven by the resulting gain of aromaticity. In favor of the assigned structure **11**, its  $^{1}$ H NMR spectrum revealed the absence of a doublet at  $\delta$  6.23, assignable to the proton on C-6 in **1**,  $^{9}$  or C-5 in **9** in their  $^{1}$ H NMR spectra. Instead, signals at  $\delta$  6.05 (C-2-H), 6.87 (C-5-H) and 10.24 (OH) were present in the  $^{1}$ H NMR spectrum of **11**. The absorption at  $v_{\text{max}}$  1733 cm $^{-1}$  in the IR spectrum of **11** is also in better agreement with the carbonyl absorption (1725–1735 cm $^{-1}$ ) of several 3-oxoindoline derivatives.  $^{10}$ 

The monoxime **1b** reacted with benzoyl-**12a** and acetyl-**12b** methylenetriphenylphosphoranes in boiling toluene, giving the dihydrobenzoxazoles **13a,b** advantageously. The structure of **13** was assigned from its molecular weight, its single carbonyl peak at  $v_{\text{max}}$  1666 (**13a**) or 1675 (**13b**); its <sup>1</sup>H NMR absorption at  $\delta \sim 3.7$  [CH-C(O)] and  $\delta \sim 11$  (NH), and its conversion by *N*-bromosuccinimide (NBS) (Scheme 3) to **14** [78%,  $v_{\text{max}} \sim 1595 \text{ cm}^{-1}$ , C=N].

Next, we studied the reaction of **1b** with allyltriphenylphosphonium bromide **15**; the obtained product is depicted in Scheme 4. The treatment of **1b** with **15** in the presence of lithium hydroxide in CHCl<sub>3</sub> yielded 6,8-di-*tert*-butyl-2-methyl-2H-1,4-benzoxazine (**19**) (48%) (and unidentified products of

$$\begin{bmatrix} Ph_3P^+-CH_2-CH=CH_2 & Br^- & \frac{LIOH}{-HBr} & Ph_3P=CH-CH=CH_2 \\ 15 & 15A \end{bmatrix} + 1bA$$

$$\begin{array}{c|c} \hline \text{CHCl}_3 \\ \hline \text{LiOH} \\ \hline \end{array} \begin{array}{c} \text{N=O} \\ + \\ \hline \end{array} \begin{array}{c} \text{CH}_3\text{-CH=CH-PPh}_3 \\ \hline \end{array} \begin{array}{c} \text{LiOH} \\ \hline \text{CHCl}_3 \\ \hline \end{array}$$

Scheme 4.

high melting points). Structure **19** was assigned from an elemental analysis and the spectral properties. Obviously, the highly reactive ylide **15A** is unlikely to react at the oxoimino nitrogen atom of **1b** and is much more likely to remove a proton from the OH group, to generate cation **17**. An attack of the oxygen anion **16** on **17** yields the ylide **18**, which would then undergo a Wittig reaction giving the final product **19**. Noteworthy, an analogous mechanism was previously reported for the reaction of vinylphosphonium salt with  $\alpha$ -imino ketones.<sup>11</sup>

In summary, the present and previous studies  $^{4g}$  clearly show that the  $\alpha$ -imino carbonyl substrates 1a,b react with alkylidenephosphoranes exclusively in the phenolic form, and not in the tautomeric o-quinone-imine structure, at least under the prevailing experimental conditions. Although the initial step in the reactions of 1b with 2 or 12 involves a nucleophilic attack by the nitroso-oxygen atom on the phosphonium center of the reagent, the consequences of the initial step vary markedly according to the nature of the  $\alpha$ -substituent of the ylidic carbon atom. Finally, the N-heterocycles prepared in the present work, might to be biologically active compounds based on known drug skeletons.

### Experimental

All of the melting points are uncorrected. IR spectra were recorded on a Perkin-Elmer spectrophotometer model 297 (Grating) using a KBr disc. The <sup>1</sup>H and <sup>13</sup>C NMR spectra were run on a Varian Gemini 200 (200 MHz) instrument using TMS as an internal reference. The mass spectra were taken at 70 eV on an MS-50 Kratos (A.E.I.) spectrometer provided with a data system. Appropriate precautions in handling moisture-sensitive compounds were observed. Light petroleum refers to the 40–60 °C fraction.

I. Reaction of 3,5-Di-*tert*-butyl-1,2-benzoquinone 1-Oxime (1b) with Ester Ylides 2a,b: (A) Preparation of Compounds 9a and 11. A stirred solution of monoxime 1b (0.8 g, 3.4 mmol) and (ethoxycarbonylmethylene)triphenylphosphorane (2a) (2.4 g, 6.8 mmol) in dry chloroform (25 mL) was boiled under reflux for 18

h. After removing the solvent, the residue was chromatographed on silica gel. Elution with hexane/CHCl<sub>3</sub> (9:1, v/v) afforded yellow crystals of ethyl 6,8-di-*tert*-butyl-2-oxo-2*H*-1,4-benzoxazine-3-acetate (**9a**) (480 mg, 41%), mp 131.5–133 °C (CH<sub>2</sub>Cl<sub>2</sub>); Anal. Found: C, 69.66; H, 7.82; N, 3.93%. Calcd for C<sub>20</sub>H<sub>27</sub>NO<sub>4</sub> (345.44): C, 69.54; H, 7.88; N, 4.05%; IR (KBr) 1763 (C=O, lactone), 1722 (C=O, ester), 1595 cm<sup>-1</sup> (C=N); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  0.95 (3H, t,  $J_{\rm HH}$  = 6.5 Hz, OC-CH<sub>3</sub>), 1.24, 1.28 (2 × 9H, 2s, C(CH<sub>3</sub>)<sub>3</sub>), 2.64 (2H, s, CH<sub>2</sub>-), 4.05 (2H, q,  $J_{\rm HH}$  = 6.5 Hz, OCH<sub>2</sub>), 6.23, 6.82 (2 × 1H, 2d,  $J_{\rm HH}$  = 4.2 Hz, Ar-*H*); <sup>13</sup>C NMR  $\delta$  18.4 (OCCH<sub>3</sub>), 27.8 (CH<sub>2</sub>-), 31.2, 32.4 [2 × C(CH<sub>3</sub>)], 34.6, 35.8 (2 × C(CH<sub>3</sub>)<sub>3</sub>), 58.3 (OCH<sub>2</sub>), 123.7, 125.5 (C-5 & C-7), 129.4 (C-4a), 133.1 (C-3), 142.0 (C-6), 144.2 (C-8), 151.5 (C-8a), 159.3 (C(O), ester), 171.5 (C-2-(O)); MS m/z (%) 345 (M<sup>+</sup>, 55), 330 (11), 300 (15), 272 (19), 271 (100), 246 (23), 186 (5).

(B) Preparation of **9b** and **11**: A stirred solution of **1b** (0.8 g, 3.4 mmol) and (methoxycarbonylmethylene)triphenylphosphorane (**2b**) (2.3 g, 6.8 mmol) in dry CHCl $_3$  (20 mL) was boiled under reflux for 15 h (TLC). The product mixture was worked up according to the above-described procedure for ylide **2a**, yielding compounds **9b** and **11**.

Elution with hexane/CHCl<sub>3</sub> (9:1, v/v) yielded yellow crystals of methyl 6,8-di-*tert*-butyl-2-oxo-1*H*-1,4-benzoxazine-3-acetate (**9b**) (490 mg, 49%), mp 138–140 °C (CH<sub>2</sub>Cl<sub>2</sub>); Anal. Found: C, 68.74; H, 7.67; N, 4.18%. Calcd for  $C_{19}H_{25}NO_4$  (331.42): C, 68.85; H, 7.60; N, 4.23%; IR (KBr) 1772 (C=O, lactone), 1718 (C=O, ester), 1598 cm<sup>-1</sup> (C=N); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.24, 1.28 (2

 $\times$  9H, 2s, C(CH<sub>3</sub>)<sub>3</sub>), 2.58 (2H, s, CH<sub>2</sub>-), 3.84 (3H, s, OCH<sub>3</sub>), 6.24, 6.79 (2 × 1H, 2d,  $J_{\rm HH}$  = 4.2 Hz, Ar-H); <sup>13</sup>C NMR  $\delta$  28.3 (CH<sub>2</sub>-), 31.4, 32.8 (2 × C(CH<sub>3</sub>)<sub>3</sub>], 34.3, 35.5 (2 × C(CH<sub>3</sub>)<sub>3</sub>), 54.5 (s, OCH<sub>3</sub>), 123.4, 125.4 (C-5 & C-7), 129.8 (C-4a), 133.3 (C-3), 141.8 (C-6), 146.4 (C-8), 150.8 (C-8a), 161.5 (C(O), ester), 173.6 (C-2-(O)); MS m/z (%) 331 (M<sup>+</sup>, 48), 316 (4), 300 (17), 272 (27), 271 (100), 246 (20), 186 (8).

Elution with hexane/CHCl<sub>3</sub> (8:2, v/v) afforded yellow leaflets of 3H-indol-3-one 11 (195 mg, 22%), mp 155–157 °C (benzene), and were shown to be identical to material prepared as described above by using 2a.

(C) A reaction between equimolar amounts of **1b** and **2a** or **2b** in CHCl<sub>3</sub> was carried out, and the reaction mixture was worked up according to the above-described procedure for **2a,b**. The product mixture gave (with **2a**) **9a** (21%) and **11** (18%) and (with **2b**) **9b** (20%) and **11** (14%). Unreacted oxime **1b** was also isolated (~25%) in each case.

(D) A reaction between **1b** and 2 molar amounts of **2a,b** was repeated in dry toluene. The reaction mixture was heated under reflux for 40 h and afforded, after the usual working up, compounds **9a,b** ( $\sim$ 28%) and **11** ( $\sim$ 17), respectively.

**II. Reactions of 1b with Keto Ylides 12a,b:** Preparation of **13a,b** and **14a,b**: A mixture of **1b** (0.8 g, 3.4 mmol) and benzoyl-**12a** or acetyl-**12b** methylenetriphenylphosphorane (4 mmol) in dry toluene (20 mL) was refluxed for 40 h. The product mixture was chromatographed on silica gel using hexane/chloroform (8:2, v/v) as the eluent to give **13a** or **13b**, respectively.

2-Benzoyl-5,7-di-*tert*-butyl-2,3-dihydrobenzoxazole (**13a**) was obtained as pale-yellow flakes (722 mg, 63%), mp 118–120 °C (cyclohexane); Anal. Found: C, 78.35; H, 8.14; N, 4.20%. Calcd for  $C_{22}H_{27}NO_2$  (337.46): C, 78.30; H, 8.06; N, 4.15%; IR (KBr) 3345 (NH), 1666 cm<sup>-1</sup> C(O)Ph; <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.23, 1.26 (2 × 9H, 2s, C(CH<sub>3</sub>)<sub>3</sub>], 3.68 (C-2-H), 6.23 (1H, d,  $J_{HH}$  = 4.2 Hz, C-4-H), 6.81 (1H, d,  $J_{HH}$  = 4.2 Hz, C-6-H), 7.35–7.77 (5H, m, Ph-H), 11.3 (1H, s, NH, exchangeable with D<sub>2</sub>O); <sup>13</sup>C NMR  $\delta$  31.2, 32.6 (2 × C(CH<sub>3</sub>)<sub>3</sub>), 34.1, 35.5 (2 × C(CH<sub>3</sub>)<sub>3</sub>), 51.3 (C-2), 123.4, 124.4 (C-4 & C-6), 142.7 (C-5), 147.4 (C-7), 195.5 (C(O), benzoyl); MS m/z (%) 337 (M<sup>+</sup>, 33), 335 (55), 307 (10), 232 (29), 230 (100), 186 (6).

2-Acetyl 5,7-di-*tert*-butyl-2,3-dihydrobenzoxazole (**13b**) was obtained as pale yellow needles (475 mg, 51%), mp 108–109 °C (pentane); Anal. Found: C, 74.23; H, 9.09; N, 5.18%. Calcd for  $C_{17}H_{25}NO_2$  (275.39): C, 74.14; H, 9.15; N, 5.08%; IR (KBr) 3330 (NH), 1675 cm<sup>-1</sup> (C(O)CH<sub>3</sub>); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.24, 1.26 (2 × 9H, 2s, C(CH<sub>3</sub>)<sub>3</sub>), 2.55 (3H, s, (O)CH<sub>3</sub>), 3.88 (C-2-H), 6.22 (1H, d,  $J_{\rm HH}$  = 4.2 Hz, C-4-H), 6.76 (1H, d,  $J_{\rm HH}$  = 4.2 Hz, C-6-H), 10.68 (1H, s, NH, exchangeable with D<sub>2</sub>O); <sup>13</sup>C NMR  $\delta$  30.8, 31.3, 32.8 (C(O)CH<sub>3</sub> & 2 × C(CH<sub>3</sub>)<sub>3</sub>), 34.6, 35.2 (2 × C(CH<sub>3</sub>)<sub>3</sub>), 48.8 (C-2),123.8, 125.9 (C-4 & C-6), 142.3 (C-5), 146.4 (C-7), 192.5 (C(O), acetyl); MS m/z (%) 275 (M<sup>+</sup>, 22), 273 (38), 258 (11), 245 (9), 230 (100), 186 (8).

Conversion of 13a,b to 14a,b: N-Bromosuccinimide (NBS) (54 mg, 0.3 mmol) and benzoyl peroxide (7 mg, 0.04 mmol) were added to a solution of 13a (100 mg, 0.3 mmol) or 13b (82 mg, 0.3 mmol) in 20 mL of dry CCl<sub>4</sub>. The mixture was refluxed for 2 h and filtered while hot. Evaporation of the solvent left a residue, which was triturated with a small amount of light petroleum to give the dehydrogenated derivative, 14a or 14b.

2-Benzoyl-5,7-di-*tert*-butylbenzoxazole (**14a**) was obtained as a pale-yellow substance (72 mg, 72%), mp 105–107 °C (pentane); Anal. Found: C, 78.65; H, 7.45; N, 4.07%. Calcd for  $C_{22}H_{25}NO_2$ 

(335.45): C, 78.77; H, 7.51; N, 4.17%; IR (KBr) 1667 (C(O)Ph), 1605 cm<sup>-1</sup> (C=N); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.23, 1.25 (2 × 9H, 2s, C(CH<sub>3</sub>)<sub>3</sub>], 6.23, 6.83 (2 × 1H, 2d,  $J_{\rm HH}$  = 4.2 Hz, C-4-H & C-6-H), 7.35–7.77 (5H, m, Ph-*H*); MS m/z (%) 335 (M<sup>+</sup>, 28), 307 (13), 231 (66), 230 (100), 185 (3).

2-Acetyl-5,7-di-*tert*-butylbenzoxazole (**14b**) was obtained as yellow crystals (57 mg, 70%), mp 89–90 °C (light petroleum); Anal. Found: C, 74.62; H, 8.58; N, 5.06%. Calcd for  $C_{17}H_{23}NO_2$  (273.38): C, 74.69; H, 8.48; N, 5.12%; IR (KBr) 1675 (C(O)Me), 1598 cm<sup>-1</sup> (C=N); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.22, 1.24 [2 × 9H, 2s, C(CH<sub>3</sub>)<sub>3</sub>], 2.57 (3H, s, (O)CH<sub>3</sub>), 6.22, 6.79 (2 × 1H, 2d,  $J_{HH}$  = 4.2 Hz, C-4-H & C-6-H); MS m/z (%) 273 (M<sup>+</sup>, 43), 258 (14), 245 (8), 230 (100), 186 (8).

III. Reaction of Monoxime 1b with Allylidenetriphenylphosphorane (16): Preparation of 19: A solution of allyltriphenylphosphonium bromide (1.4 g, 3.6 mmol) and the monoxime **1b** (0.8 g, 3.4 mmol) in chloroform (40 mL) was stirred by a magnetic stirrer. Freshly prepared aqueous lithium hydroxide (0.5 M, 15 mL, 1 M = 1 mol dm<sup>-3</sup>) was added in one portion to the mixture, and the two- phase system was stirred at room temperature for 1 h, then refluxed for 4 h. The product mixture was then extracted with CHCl<sub>3</sub> (2 × 50 mL) and dried; the solvent was removed under reduced pressure. The residue was chromatographed on silica gel using hexane/CHCl<sub>3</sub> 7:3, v/v to give 6,8-di-tert-butyl-2-methyl-2H-1,4-benzoxazine (19) (425 mg, 48%), mp 174–175 °C (acetonitrile); Anal. Found: C, 78.77; H, 9.67; N, 5.32%. Calcd for C<sub>17</sub>H<sub>25</sub>NO (259.39): C, 78.71; H, 9.71; N, 5.40%; IR (KBr) 1600 cm<sup>-1</sup> (C=N); <sup>1</sup>H NMR (CDCl<sub>3</sub>)  $\delta$  1.24, 1.28 (2 × 9H, 2s,  $C(CH_3)_3$ , 1.68 (3H, d,  $J_{HH} = 8.2$  Hz, C-2-CH<sub>3</sub>), 3.84 (1H, d of q,  $J_{\rm HH} = 5.8$  Hz, C-2-H), 6.25 (1H, d,  $J_{\rm HH} = 2.4$  Hz, C-5-H), 6.87– 6.95 (2H, m, C-3-H & C-7-H);  $^{13}$ C NMR  $\delta$  22.2 (C-2-CH<sub>3</sub>), 31.1,  $32.4 (2 \times C(CH_3)_3), 34.3, 35.3 (2 \times C(CH_3)_3), 44.7 (C-2), 122.6,$ 123.3, 125.2 (C-3, C-5 & C-7), 129.8 (C-4a), 142.4 (C-6), 146.6 (C-8), 150.4 (C-8a); MS m/z (%) 259 (M<sup>+</sup>, 57), 244 (100), 228 (21), 226 (11), 199 (7), 186 (10).

In the next fractions several polymeric unidentified products with mp > 300  $^{\circ}$ C were eluted.

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